# The thermal and thermomechanical properties of Ti nanocomposites based hydroxyapatite



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### Introduction

Hydroxyapatite Ca<sub>10</sub>(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>, HA, is the principal inorganic component of bones and teeth. Synthetic HAP is mainly applied in repair and replacements of hard tissues, presenting excellent biocompatibility. However ceramics HA can be easily decomposed at a temperature depending on the conditions of synthesis, which results in poor sintering characteristics and mechanical properties [1]. The addition of TiH2 on hydroxyapatite followed by calcination can lead to the formation of a composite with improved mechanical properties without affecting the biocompatibility of the material. Relations between thermal stability and mechanical properties of pure HA and HA+Ti composite ceramics were investigated by thermogravimetric analysis, differential scanning calorimetry and thermomechanical analysis.

## Sample preparation

For synthesis of nanocomposites based hydroxyapatite, HA commercial and Ti powder were used. HA commercial (<200 nm, type Aldrich) was calcinated at 900°C for 1 hour in air. The reinforcement is micrometric Ti powder particles (~100 µm, water atomised). The weight ratio between the components is 75% HAP and 25% Ti. The mixture was homogenized for 20 min. in a planetary ball mill, Fritsch - Pulverisette 6, n = 200 rpm, in liquid (ethanol). Then the composite mixture was dried at 200°C in air, and then remixed in the planetary ball mill for 1 min.[2].

### Samples characterization

Samples structural characterized by X-ray diffraction. Fourier transform infrared spectra were measured in the range of 4000-400cm<sup>-1</sup>. spectrophotometer. The specific surface area of the samples were measured by the nitrogen adsorption technique (NOVA 2200e).

Tab. 1. Samples characteristics from XRD and BET

	Crystallite	BET surface
	size (nm)	area (m <sup>2</sup> g <sup>-1</sup> )
Ca <sub>5</sub> (PO4) <sub>3</sub> (OH) - HA	46	19.192
-		
Ca <sub>5</sub> (PO4) <sub>3</sub> (OH) – HA	145	9.504
Ca <sub>5</sub> (PO4) <sub>3</sub> (OH) - HA:75%	-	9.213
Ti: 25%		
	Ca <sub>5</sub> ( PO4 ) <sub>3</sub> ( OH ) – HA Ca <sub>5</sub> ( PO4 ) <sub>3</sub> ( OH ) – HA:75%	Ca <sub>5</sub> (PO4) <sub>3</sub> (OH) - HA 46  Ca <sub>5</sub> (PO4) <sub>3</sub> (OH) - HA 145  Ca <sub>5</sub> (PO4) <sub>3</sub> (OH) - HA:75% -

The OH libration bands at 630 cm-1 are visible on HA calcinated and HA+Ti spectra and the OH stretching bands at 3571 cm<sup>-1</sup> are found on HA commercial and HA calcinated spectra. The characteristic bands due to PO<sub>4</sub>3- ions are clearly visible at 568, 601, 962, 1026, 1092 cm-1 wave numbers.

## Thermal analysis

The thermodynamic stability of the HA and HA-Ti ceramics in the temperature range from 40 to 1400°C was investigated by using a SETSYS Evolution Setaram DSC/TG. The experiments were done on powder samples, with a heating rate of 10°C/min, under Ar flow.

The thermomechanical behavior was investigated by using a SETARAM Thermo Mechanical Analyzer (SETSYS Evolution TMA). The experiments were done on ceramic samples (pellets form), at a heating rate 5°C/min and under Ar flow.

T/°C Fig. 3 DSC data in the 40 - 1400 °C temperature range

HA commercial HA calcinated

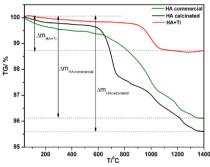


Fig. 2. TG data in the  $40-1400\,^{\circ}\text{C}\,$  temperature range

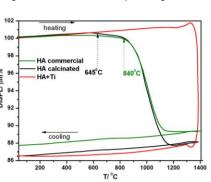


Fig. 4. TMA data in the 40 - 1400 °C temperature range

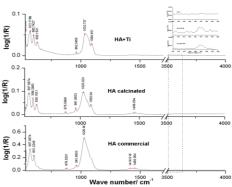
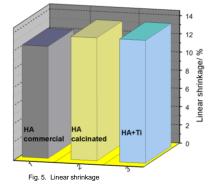


Fig. 1. IR spectra of the samples

Sample	Total mass loss (%)
HA commercial	-3,9%
HA calcinated	-4,4%
HA+Ti	-1.3%

Tab. 2. Total mass loss in the 40 – 1400 °C temperature range

In the temperature range from 40 to 800°C an insignificant mass loss observed on TG curve for HA+Ti sample is in good agreement with the absence of OH stretching bands at 3571 cm<sup>-1</sup> corresponding adsorbed water in the FTIR spectrum.



• The comparison among DSC heating curves of composite HA+Ti and pure HA (commercial and calcinated) indicates that the thermal stability domain of HA+Ti composite is shifted to higher temperatures. The HA decomposition occurs at higher temperature for composite than pure HA (Fig. 3).

• The Ti addition on HA do not affect the linear shrinkage of HA+Ti composite (Fig. 4, 5).

1200

1400

### Conclusions

- ❖The influence of Ti addition on dehidroxylation, decomposition and sintering temperature of HA+Ti sample is evidenced.
- ❖HA+Ti Nanocomposite is characterized by a higher thermal stability domain.
- The nanocomposite HA+Ti has improved mechanical properties being an advantage for biomedical applications.

#### References

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### Acknowledgements

The financial support of the INFRANANOCHEM Nr. 19/01.03.2009 and National Program II—Partnership ctr. 72-184 grant are acknowledged.